

# THE MECHANISM OF THE INHIBITION BY CARBONIC ACID OF THE SMOOTH MUSCLE CONTRACTION PRODUCED BY HISTAMINE AND OXYTOCIN

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Carbonic acid inhibited the stimulating action of histamine on guinea-pig intestine and of oxytocin on the uterus of many animal species. This effect was relatively specific. Under the same conditions the actions of acetylcholine, KCl, adrenaline (on rabbit uterus), and ergot alkaloids were not inhibited. The electrical excitability of smooth muscle was similarly not affected. The inhibitory action of carbonic acid is directly proportional to its concentration in the medium in which the isolated organ is maintained. The study of the activity of histamine and oxytocin at different pH in a medium free of NaHCO<sub>3</sub> and buffered with a mixture of sodium maleate and maleic acid suggested that the inhibitory action exerted by carbonic acid was specific and independent of the simultaneous modifications of the hydrogen ion concentration. The mechanism of these phenomena is discussed.

Evidence has been provided recently that carbon dioxide inhibits the excitability of peripheral nerves by a specific action (Lorente de N6, 1947; Monnier, 1952; Coraboeuf, 1951, 1954). On the other hand, as early as 1923 Lovatt-Evans and Underhill had shown that CO<sub>2</sub> exerts an inhibitory action on the contractions of isolated intestine produced by different substances. Later, Garan (1938) showed that the activity of histamine on isolated guinea-pig intestine was a function of the partial pressure of CO<sub>2</sub> and attributed the effect to the formation of a carbamate compound, supposedly inactive, between histamine and CO<sub>2</sub>. The conclusions of Garan (1938) were disproved by Kiese (1940), who also observed that CO<sub>2</sub> diminished the response of the intestine to histamine, although the action of acetylcholine was but little influenced and that of pilocarpine was enhanced. By studying the relationship between pH and the activity of histamine, Kiese (1940) concluded that the change in the polarity of the histamine molecule by modifications of pH could not satisfactorily explain the effect observed.

More recently, Halpern (1956a and b) studied the action of CO<sub>2</sub> on the contractions of the isolated guinea-pig ileum produced by histamine, acetylcholine, KCl and the antigen-antibody reaction. An inhibitory effect was found on

contractions produced by histamine and in anaphylactic reactions, but under the same conditions the effect of acetylcholine remained unchanged or was increased. The effect of CO<sub>2</sub> was shown to be completely reversible. It could not be attributed to tissue anoxia, because it was produced by gas mixtures containing 20% CO<sub>2</sub> and 80% O<sub>2</sub>: the partial pressure of oxygen was therefore higher than when air was used to oxygenate the medium.

Halpern, Mayer, and Bugnard (1956) have demonstrated that CO<sub>2</sub> also inhibits the action of oxytocin on the uterus of the rabbit, guinea-pig and rat. Here, too, the effect was specific for oxytocin and did not apply to the action of other substances which cause the uterus to contract.

The aim of the present work was to discover how carbon dioxide inhibits the action of histamine and oxytocin on smooth muscle. The results suggest that carbon dioxide modifies the reactivity of smooth muscle to these substances by a specific action which is not dependent upon the changes it produces in the pH of the medium.

## METHODS

The investigations were performed either with isolated guinea-pig intestine or isolated uteri from guinea-pigs, rabbits, and rats. The organs were taken

immediately after killing the animals by carotid artery section. Strips of 2 to 3 cm. were cut off and immersed in oxygenated Tyrode solution at 37° containing the following ingredients in 100 ml.: NaCl 0.8 g.; CaCl<sub>2</sub> 0.01 g.; KCl 0.02 g.; MgCl<sub>2</sub> 0.01 g.; NaH<sub>2</sub>PO<sub>4</sub> 0.005 g.; NaHCO<sub>3</sub> 0.1 g.; dextrose 0.1 g.; atropine 3 μg. When the concentration of NaHCO<sub>3</sub> was changed, a modification was made in the NaCl concentration to maintain the same molarity of the solution. In one series of experiments, a solution without NaHCO<sub>3</sub> was used and the pH was adjusted to the desired value with a buffer of sodium maleate and maleic acid. The composition of Tyrode solution was modified in this instance by replacing the NaHCO<sub>3</sub> by 0.07 g. of maleic acid (6 mM). The pH was then adjusted with N-NaOH (approximately 1.1 ml./l. to obtain pH=7.5). The solutions of histamine dihydrochloride, acetylcholine chloride, oxytocin, etc., were prepared in Tyrode solution at pH 7.4. The volume added to the bath was 0.1 to 0.5 ml. The total volume of the vessel in which the organ was immersed was 50 ml. The contractions were recorded on smoked paper, by frontal inscription, employing a magnification of about 20 times.

**Electrical Stimulation.**—The technique described by Paton (1956) was used, using the electronic stimulator of Ead (1951). The anode was platinum wire, 4 to 5 cm. long, introduced in the lumen of the intestine. The cathode was a similar wire in the solution. Stimulation was coaxial. A diagram of the apparatus is seen in Fig. 1.

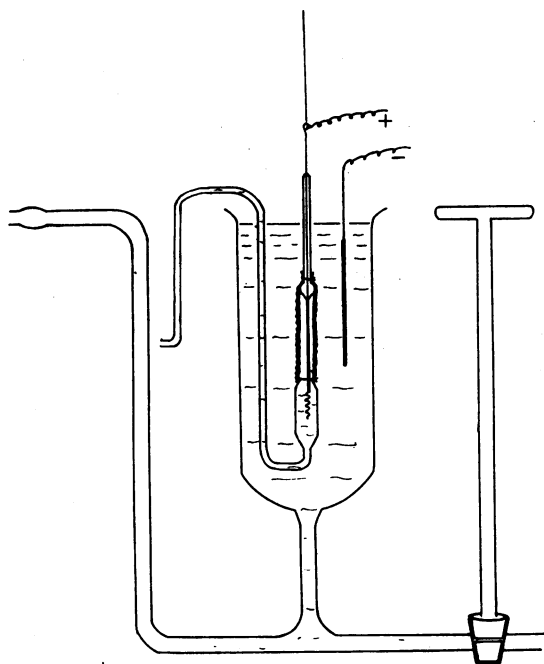


FIG. 1.—Diagram of apparatus for electrical coaxial stimulation of a smooth muscle tube.

Rectangular pulses of 20 msec. duration every 10 sec. were used to stimulate the gut. 20 V usually produced supramaximal stimulation.

**pH Measurement.**—The pH of the solution was measured instantaneously with a laboratory type pH meter (Cambridge Instrument Co.). The glass and calomel electrodes were immersed in the vessel in which the organ was studied.

**Introduction of Gas Mixtures.**—Air, oxygen, CO<sub>2</sub>, or mixtures of CO<sub>2</sub> and O<sub>2</sub> were bubbled through the solution in each experiment. With air or oxygen, the pH of the Tyrode solution, initially about 7.5, slowly increased to 8.2 or even 8.5. A mixture rich in CO<sub>2</sub> produced a diminution of the pH. A typical experiment is illustrated in Fig. 2. By simultaneously bubbling air and a mixture of CO<sub>2</sub> and O<sub>2</sub>, at different rates, the desired pH could be maintained for long enough to carry out the experiments. The H<sub>2</sub>CO<sub>3</sub> content of the solution was always calculated from the measured value of pH according to the Henderson-Hasselbach equation:

$$pH = pK + \log \frac{(\text{HCO}_3^-)}{(\text{H}_2\text{CO}_3)}$$

where pK is the negative logarithm of the dissociation constant of carbonic acid. The value of pK, at 37°, is 6.31. In our calculations no allowance was made for activity coefficients and (HCO<sub>3</sub><sup>-</sup>) was assumed to be equal to the NaHCO<sub>3</sub> content of the solution employed.

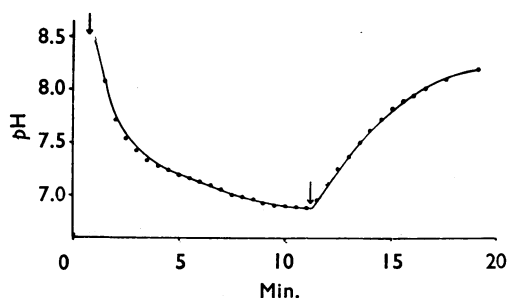


FIG. 2.—Modification of the pH of the normal Tyrode solution produced by the bubbling with 20% CO<sub>2</sub> from first to second arrow. At second arrow bubbling with air was commenced.

## RESULTS

### *Relationship Between H<sub>2</sub>CO<sub>3</sub> Concentration and Reactivity of the Isolated Guinea-pig Intestine to Histamine*

Two methods were used to evaluate the effect of the H<sub>2</sub>CO<sub>3</sub> on the action of histamine. With the first method, the height of the contraction of the isolated guinea-pig intestine produced by a fixed dose of histamine at different H<sub>2</sub>CO<sub>3</sub> concentrations was determined. The results are shown in Fig. 3.

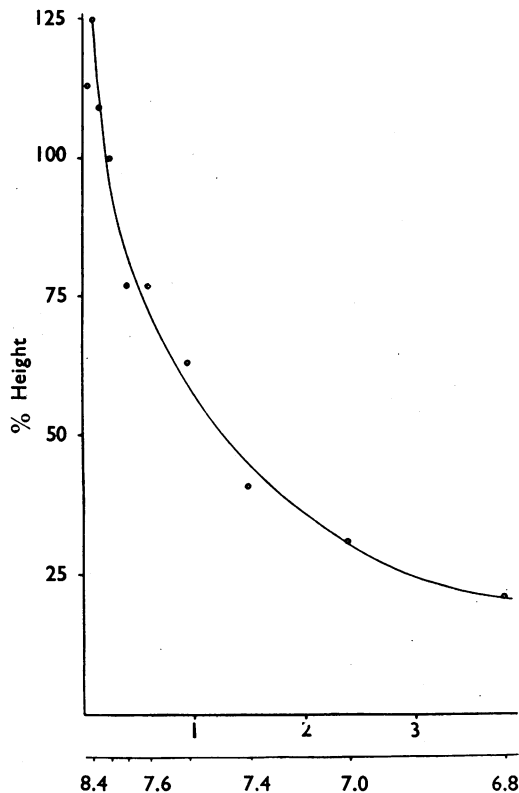


FIG. 3.—The height of the contraction of the isolated guinea-pig intestine preparation produced by a fixed quantity of histamine at various H<sub>2</sub>CO<sub>3</sub> concentration in normal Tyrode solution expressed as % of that at pH 8.0 to 8.1 given the arbitrary value of 100. The pH values are indicated on the lower abscissa, while the H<sub>2</sub>CO<sub>3</sub> concentration (mM/l.) is shown on the upper abscissa. Each point was based on at least 5 determinations with each of 4 strips of muscle.

In comparing the different contractions, the height of the one taking place at pH 8.0 to 8.1 was taken as 100%, and the heights at other pH values were referred to it. In every case, the same dose of histamine was used through the entire experiment and it was chosen so that vigorous but submaximal contractions were obtained. The dose of histamine base varied between 0.015 and 0.030  $\mu$ g./ml., depending on the particular sensitivity of each intestinal strip.

The results show that the activity of histamine was diminished as the H<sub>2</sub>CO<sub>3</sub> concentration was increased (Fig. 3). With a mixture containing 5% CO<sub>2</sub> and 95% O<sub>2</sub>, the pH attained was 7.3 and the action of histamine was diminished by about 50% of that at pH 8.0. With a mixture containing 20% CO<sub>2</sub>, equilibrium was reached at pH 6.8 and the diminution of activity was 80%.

It is impossible to measure precisely the degree of inhibition by this method because there was no direct relation between the height of the contraction and the concentration of histamine. Furthermore, at a pH of less than 6.8, the activity was so low that measurements could not be continued. To overcome these difficulties the second method was employed. This required the determination of the dose of histamine which produced the same response of the isolated intestine at different H<sub>2</sub>CO<sub>3</sub> concentrations. The results in Fig. 4 show that, when equiactive doses of histamine are plotted against the H<sub>2</sub>CO<sub>3</sub> concentration, a straight line was obtained. As the quantity of histamine necessary to produce the same effect was the reciprocal of its activity, it can be deduced from the observed relationship that the activity of the histamine was inversely proportional to the concentration of H<sub>2</sub>CO<sub>3</sub>.

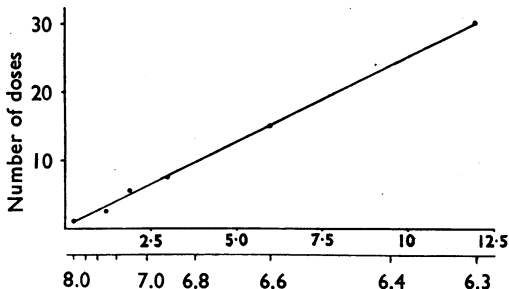


FIG. 4.—Dose of histamine (expressed in terms of the multiple of the dose required at pH 8.0) producing the same response of the isolated guinea-pig intestine preparation at various H<sub>2</sub>CO<sub>3</sub> concentration, in normal Tyrode solution.

#### Role of pH

As the action of histamine on smooth muscle was modified by the H<sub>2</sub>CO<sub>3</sub> content of the Tyrode solution, it was necessary to find whether this effect was due to changes in the pH or was produced specifically by H<sub>2</sub>CO<sub>3</sub>.

It is impossible to modify the H<sub>2</sub>CO<sub>3</sub> concentration of Tyrode solution without simultaneously changing the pH, all the other conditions remaining the same. We approached this problem in two ways: (a) by modifying the NaHCO<sub>3</sub> of Tyrode solution; (b) by employing a buffer of different composition.

*Influence of NaHCO<sub>3</sub> Concentration on the Reactivity of the Isolated Guinea-pig Intestine to Histamine and Acetylcholine.*—By using Tyrode solutions which contained between 0.25 and 10 g./l. of NaHCO<sub>3</sub>, it was possible to obtain a wide variation in pH with the same concentration of H<sub>2</sub>CO<sub>3</sub>. When air is bubbled through solutions containing 5 or 10 g./l. NaHCO<sub>3</sub>, the pH reaches

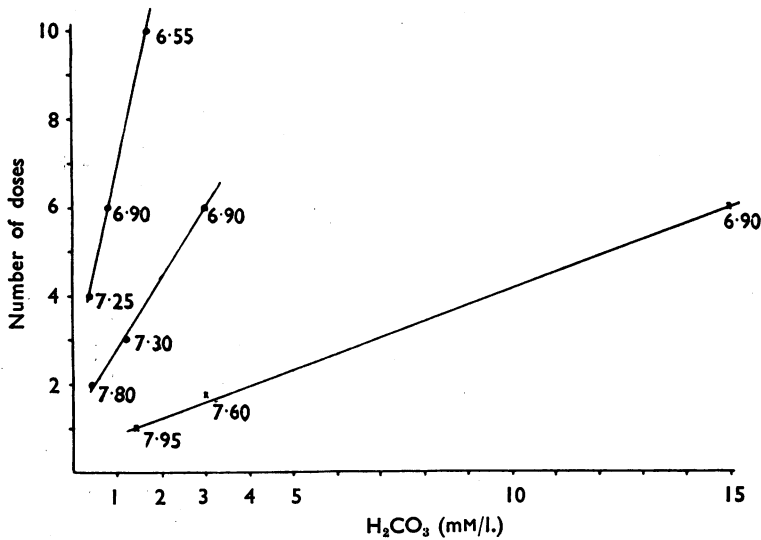


FIG. 5.—Dose of histamine (see Fig. 4) producing the same response of the isolated guinea-pig intestine preparation at various  $H_2CO_3$  concentration and at various  $NaHCO_3$  concentration. Upper curve  $NaHCO_3$  0.25 g./l.; middle curve  $NaHCO_3$  1 g./l.; lower curve  $NaHCO_3$  5 g./l. The numerals on the graph indicate the pH value at each point.

9 or more and the medium becomes quite unfavourable for the organ.

The results in Fig. 5 show that the effect of histamine diminished with increasing  $H_2CO_3$  concentration at the same  $NaHCO_3$  concentration and increased with increasing  $NaHCO_3$  concentration at the same  $H_2CO_3$  concentration. In addition, the effect of histamine on the guinea-pig intestine remained practically the same at constant pH and independent of the actual concentration of  $NaHCO_3$  and  $H_2CO_3$  (Fig. 6).

It was therefore impossible to decide, in this way, the exact rôle of pH.

*Experiments with a Different Buffer System.*—Many solutions were tried to find a convenient medium, free from  $NaHCO_3$ . Finally a solution containing maleic acid and sodium maleate proved quite suitable and, in it, organs responded regularly and reversibly for the periods of time essential for quantitative study. This was not so with solutions buffered with substances such as phosphate or citrate, which produced more or less irreversible modifications in the responses of the tissues.

Fig. 7 shows the results obtained when the activity of histamine was examined in the maleic acid/sodium maleate solution at different pH values, and, for comparison, the observations upon histamine activity in normal Tyrode solution of which the pH was modified by  $CO_2$ .

It appeared that the activity of histamine was not greatly affected by changes in pH, unless these were produced by increasing the  $H_2CO_3$  concentration. In fact, at pH 6.8 the action of histamine was practically unchanged in the maleate medium, while it was very much reduced in normal Tyrode solution. It may be concluded, therefore, that carbonic acid exerts some more or less specific action.

*Influence of  $H_2CO_3$  on the Smooth Muscle Contractions Produced by Electrical Stimulation or by Substances Other than Histamine*

It was of interest to investigate the specificity of the action of  $H_2CO_3$  on the response to histamine. The results were as follows: The action of a standard dose of acetylcholine in guinea-pig intestine was un-

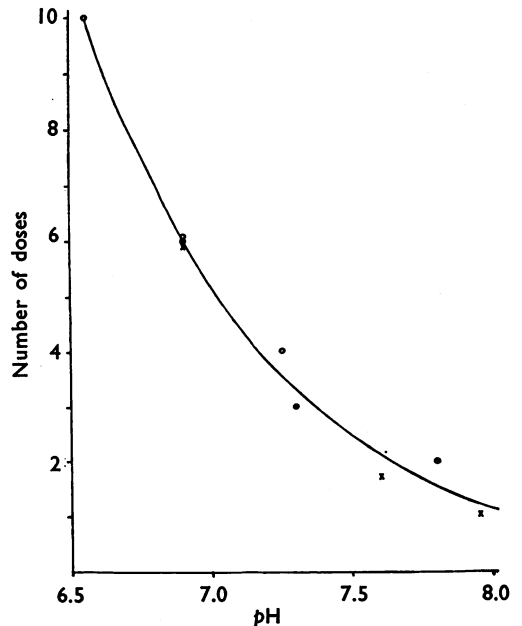


FIG. 6.—Dose of histamine (see Fig. 4) producing the same response of the isolated guinea-pig intestine preparation at different pH values, using Tyrode solution with different concentrations of  $NaHCO_3$  (O=0.25 g./l.; ●=1 g./l.; X=5 g./l.).

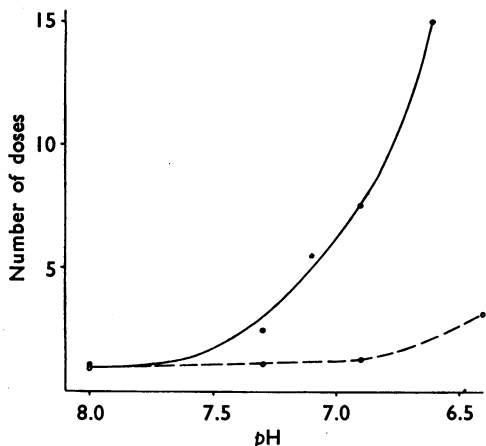


FIG. 7.—Dose of histamine (see Fig. 4) producing the same response of the isolated guinea-pig intestine preparation at different pH values. Full line, in normal Tyrode solution (bicarbonate buffer); broken line, in Tyrode solution buffered with sodium maleate and maleic acid.

affected or enhanced by the H<sub>2</sub>CO<sub>3</sub> within the limits studied (Fig. 8). The action of KCl was not modified between pH 8.1 and 6.9. The contraction of the intestine produced by electrical excitation was not modified by H<sub>2</sub>CO<sub>3</sub> between pH 8.1 and 6.9 (Fig. 9).

*Influence of H<sub>2</sub>CO<sub>3</sub> Concentration on the Response of the Isolated Uterus to Oxytocin*

Halpern *et al.* (1956) showed that H<sub>2</sub>CO<sub>3</sub> can inhibit the action of oxytocin on the isolated uterus as illustrated in Fig. 10. Experiments with KCl, adrenaline (rabbit uterus) and methyl-ergometrine (Methergin) showed that H<sub>2</sub>CO<sub>3</sub> only modified the oxytocin response (Fig. 11). The relationship of the activity of oxytocin to pH

showed that it was greatly inhibited at pH 6.8 in Tyrode solution (bicarbonate buffer) but that it was not changed at the same pH in the maleate buffered medium (Fig. 12). The observations upon histamine and oxytocin thus resemble each other.

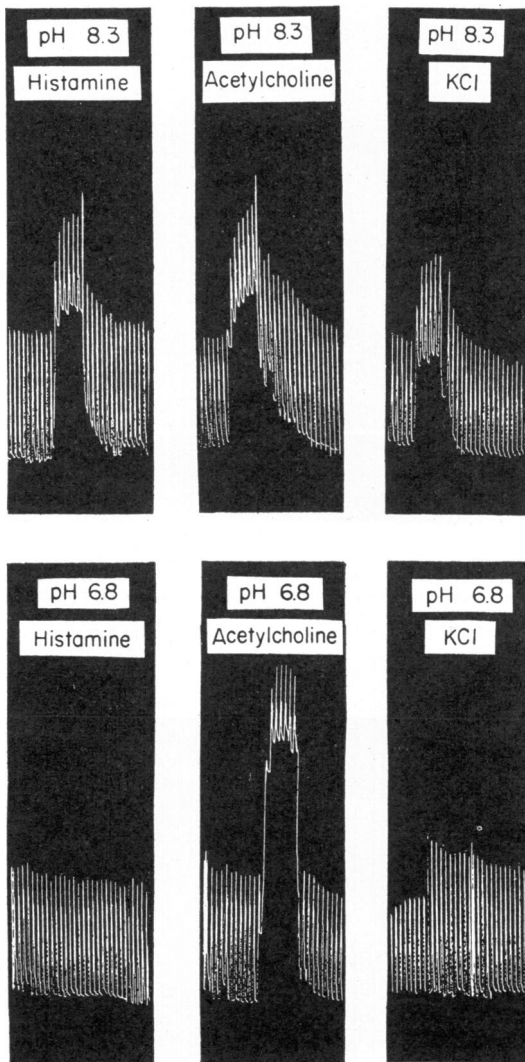


FIG. 9.—Contractions of the isolated guinea-pig ileum preparation in response to periodic electric stimulation. Pulse width, 20 msec.; frequency, 6 min.; 20 V. Upper records: effects of histamine (1.5 × 10<sup>-8</sup>), of acetylcholine (0.5 × 10<sup>-8</sup>) and KCl (1 × 10<sup>-3</sup>) on the responses of the intestine in standard Tyrode solution (buffer, NaHCO<sub>3</sub> 1 g./l.), bubbled with pure oxygen. Lower records: effects of identical concentrations of the same substances on the intestine in the same Tyrode solution, the pH of which has been lowered to 6.8 by bubbling with a gaseous mixture containing 75% O<sub>2</sub> and 25% CO<sub>2</sub>.

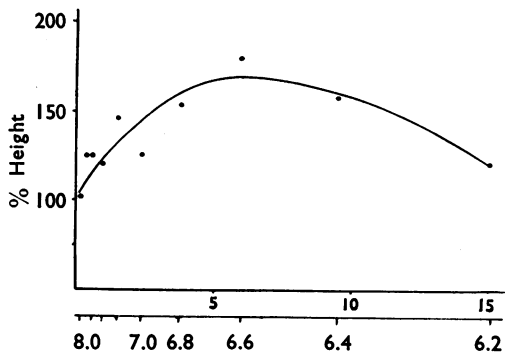


FIG. 8.—Height of the contraction (see Fig. 3) of the isolated guinea-pig intestine preparation produced by a fixed quantity of acetylcholine at different H<sub>2</sub>CO<sub>3</sub> concentration, in normal Tyrode solution. Upper abscissa, H<sub>2</sub>CO<sub>3</sub> mm/l.; lower abscissa, pH.

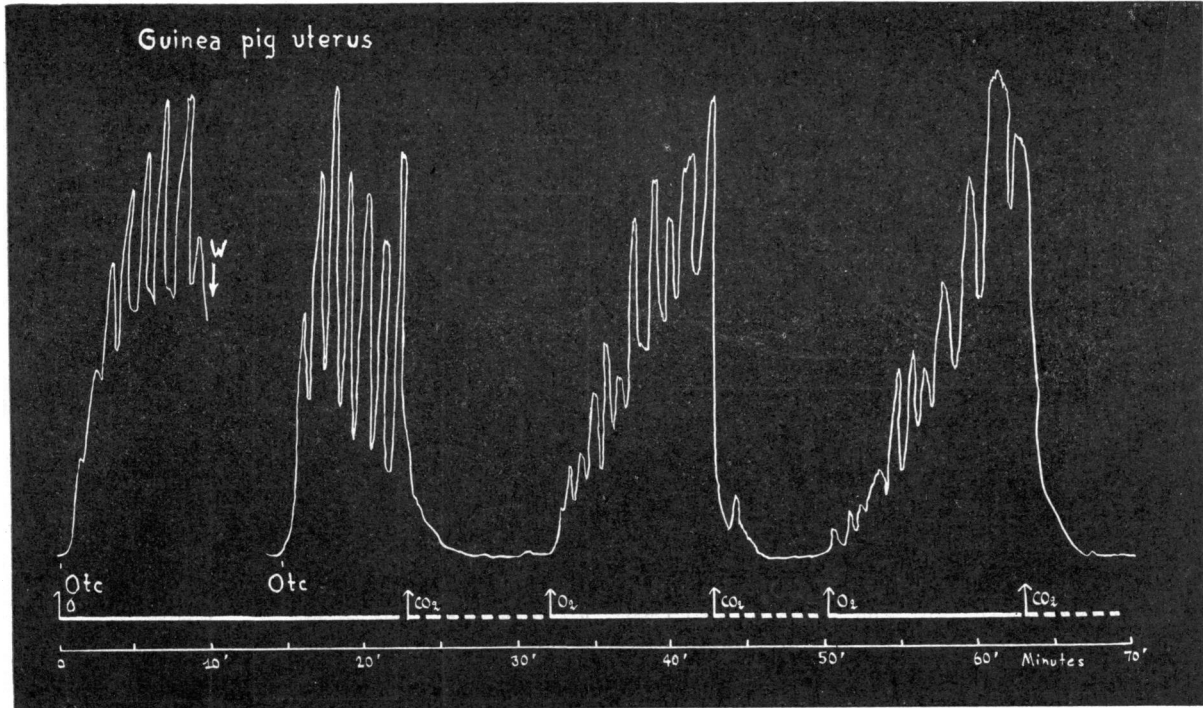


FIG. 10.—Isolated guinea-pig uterus preparation in standard Tyrode solution (buffer,  $\text{NaHCO}_3$  1 g./l.). From left to right: Effect of a standard dose of oxytocin ( $1 \times 10^{-4}$  I.U.) on the uterus in Tyrode solution oxygenated with pure oxygen. After washing, addition of the same dose of hormone produces similar effect. Without removing the hormone, a gaseous mixture containing 75%  $\text{O}_2$  and 25%  $\text{CO}_2$  was substituted for oxygen at the point marked  $\text{CO}_2$ . At the point  $\text{O}_2$ , pure oxygen was substituted for the gaseous mixture. The same procedure is repeated twice. Otc=Oxytocin; W=wash. Full line=pure oxygen; broken line=gaseous mixture containing 75%  $\text{O}_2$  and 25%  $\text{CO}_2$ . Time, 5 min.

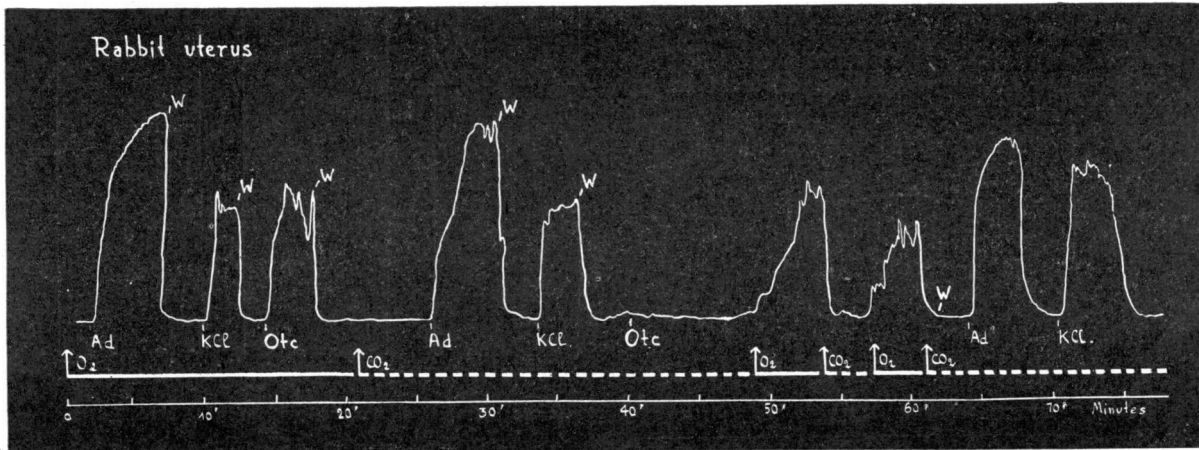


FIG. 11.—Isolated rabbit uterus preparation in standard Tyrode solution (buffer,  $\text{NaHCO}_3$  1 g./l.). From left to right: Effects of a standard dose of adrenaline ( $2 \times 10^{-6}$ ), KCl ( $5 \times 10^{-4}$ ) and of oxytocin ( $1 \times 10^{-4}$  I.U.) on the uterus in Tyrode solution oxygenated with pure oxygen (full line). After washing, a gaseous mixture containing 75%  $\text{O}_2$  and 25%  $\text{CO}_2$  was substituted at the point  $\text{CO}_2$  and the responses to the drugs again tested. Without washing out the oxytocin, pure oxygen was substituted for the gaseous mixture at  $\text{O}_2$ . Ad=adrenaline; Otc=oxytocin; W=wash. Full line=pure oxygen; broken line=gaseous mixture containing 75%  $\text{O}_2$  and 25%  $\text{CO}_2$ . Time, 5 min.

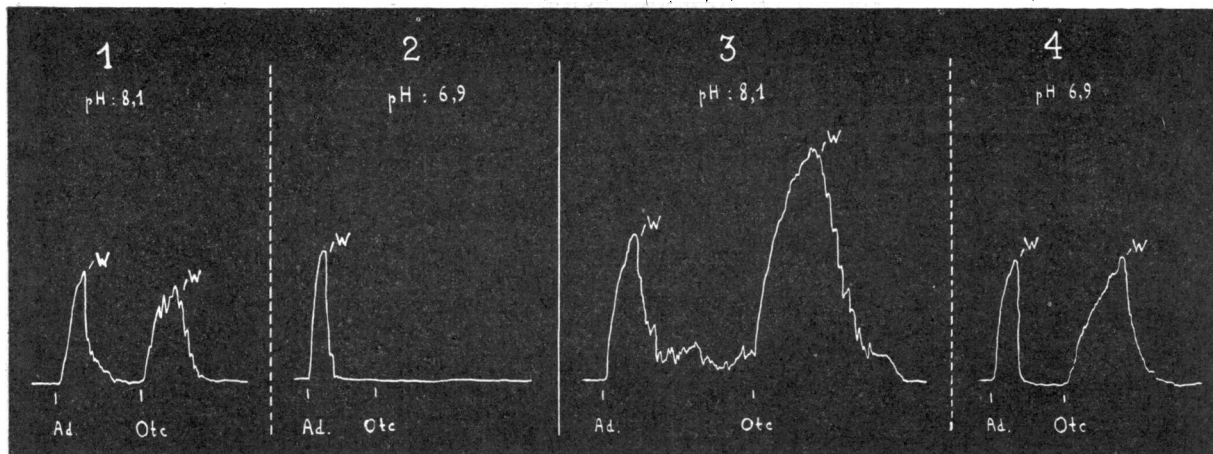


FIG. 12.—Isolated rabbit uterus horn preparation. On the left side (1 and 2) in standard Tyrode solution (buffer, NaHCO<sub>3</sub> 1 g./l.); on the right side (3 and 4) in Tyrode solution buffered with maleic acid/sodium maleate. Ad=adrenaline; Otc=oxytocin; W=wash. In 1, effects of a standard dose of adrenaline ( $2 \times 10^{-6}$ ) and of oxytocin ( $2 \times 10^{-4}$  I.U.) in normal Tyrode solution oxygenated with pure oxygen. In 2, a gaseous mixture containing 75% O<sub>2</sub> and 25% CO<sub>2</sub> was substituted for oxygen and the pH fell to 6.9. Note response to adrenaline but the absence of response to oxytocin in the same dosages as in 1. In 3, Tyrode solution buffered with maleic acid/sodium maleate. Effects of the same doses of adrenaline and of oxytocin. In 4, the pH of the Tyrode solution has been lowered to 6.9 by change of the buffer constituents (maleic acid/sodium maleate) while the medium is oxygenated with pure oxygen and the effects of the same doses of adrenaline and oxytocin are shown.

#### DISCUSSION

The results indicate that carbonic acid inhibited the action of histamine and oxytocin on smooth muscle. As has been previously reported, the effect was not due to tissue anoxia. It is crucial to know if the inhibition was produced primarily by reduction in the pH of the medium due to modifications in the H<sub>2</sub>CO<sub>3</sub> concentration. The introduction of CO<sub>2</sub> into Tyrode solution, buffered with NaHCO<sub>3</sub>, decreased the pH, and it seems logical to relate this fact to the inhibitory action. This supposition is supported by the results of experiments at various NaHCO<sub>3</sub> concentrations.

But this hypothesis is invalidated by the fact that when the bicarbonate buffer was replaced by a mixture of sodium maleate and maleic acid, the activity of histamine remained the same in spite of the variations of pH. In the presence of H<sub>2</sub>CO<sub>3</sub>, the action of histamine is diminished about 7 times at pH 6.8 in comparison with that at pH 8.1, yet it is hardly affected in the maleate medium (Fig. 8). This leads to the conclusions that the inhibition of histamine activity was independent of pH and that it was related to the presence of carbonic acid and its salt. The same appears to be true for oxytocin. In some experiments (unpublished observations), similar results were obtained employing a bicarbonate-free Tyrode solution buffered with phosphate.

We can then postulate that the inhibition of the action of histamine and oxytocin is not due to the decrease of pH, but that it is determined by a qualitative effect of carbonic acid. What is the nature of the effect? The formation of an inactive compound between histamine and H<sub>2</sub>CO<sub>3</sub> seems unlikely. Possibly H<sub>2</sub>CO<sub>3</sub> modifies the sensitivity of the specific cellular receptors for certain substances.

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